

# OPTIMUM CONDITIONS OF A CARBON DIOXIDE TRANSCRITICAL POWER CYCLE FROM LOW TEMPERATURE HEAT SOURCE FOR POWER GENERATION.

Jhon Fredy Vélez Jaramillo.  
CARTIF Technological Center; Valladolid (Spain).



Delft, The Netherlands

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**José J. Segovia<sup>(1)</sup>, M. Carmen Martín<sup>(1)</sup>, Gregorio Antolin<sup>(2)</sup>, Cecilia Sanz<sup>(2)</sup>,  
Farid Chejne<sup>(3)</sup>.**

**(1) TERMOCAL; University of Valladolid; Valladolid (Spain).**

**(2) CARTIF Technological Center; Valladolid (Spain).**

**(3) TAYEA, National University of Colombia, Medellín (Colombia).**



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# Introduction

- ✗ Worldwide energy demand is continuously increasing.

- ✗ More than 50% of the fuel we use is waste.

But.....

- ✗ Conventional steam power cycles cannot give a better performance to recover low-grade waste heat.

- ✗ Organic Rankine Cycles show the known pinching problem.

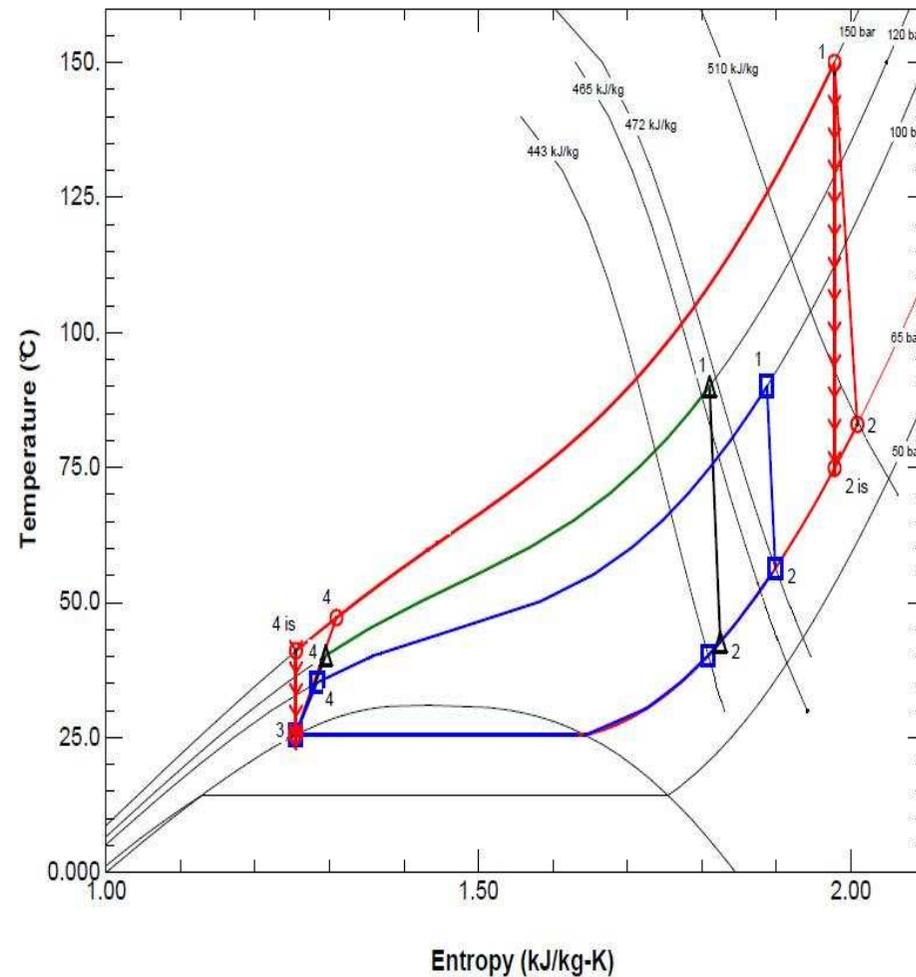
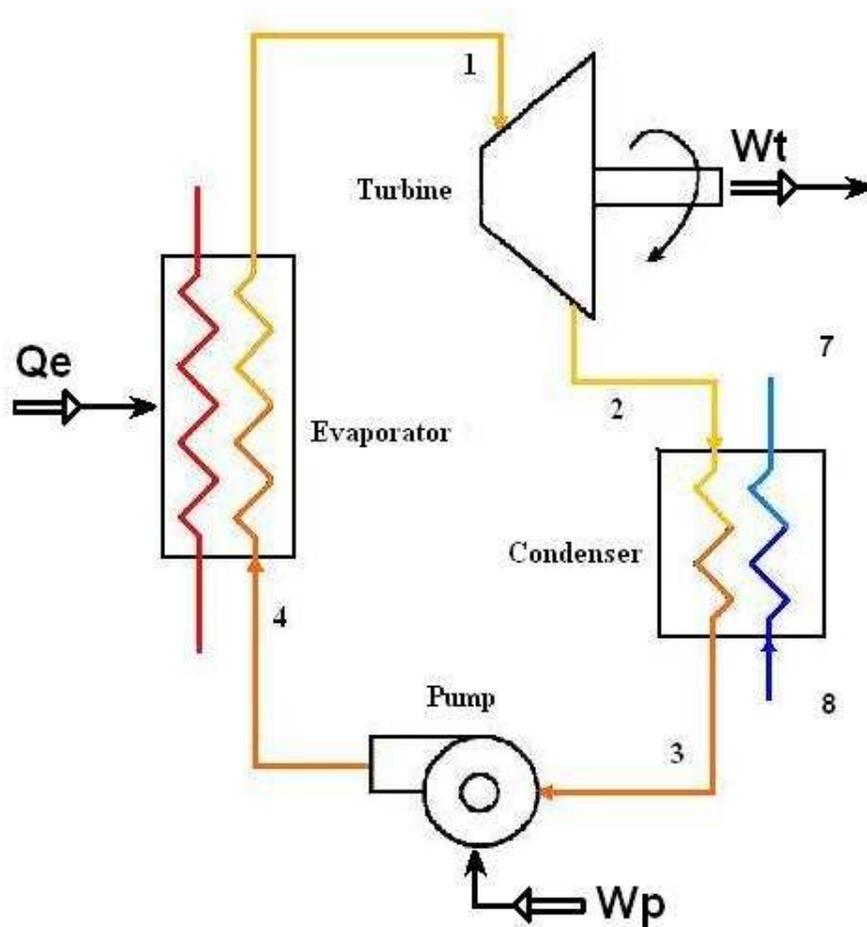
So.....

- ✓ **Use of carbon dioxide in transcritical conditions in a Rankine cycles can solve both problems.**

# Description of the CDTPC

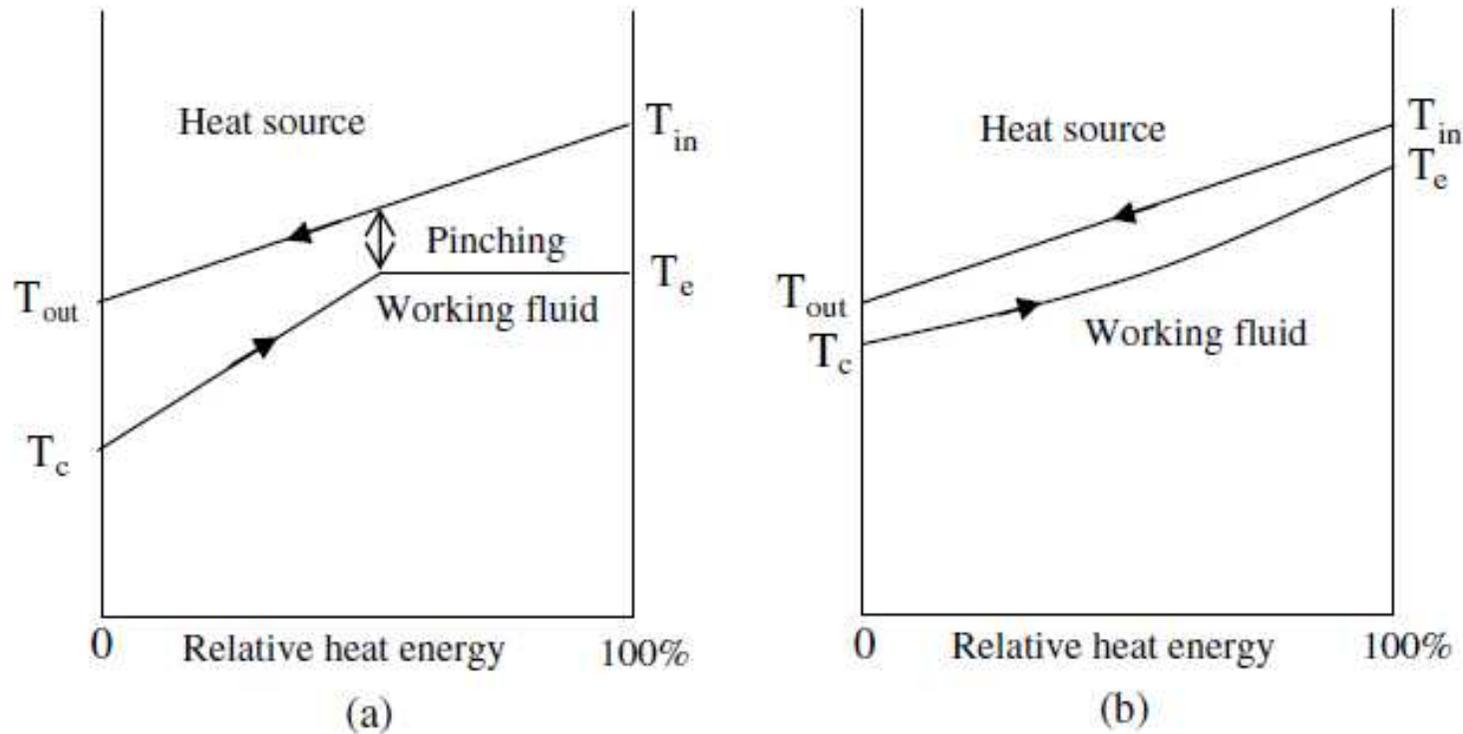
- Rankine cycle where the working fluid ( $\text{CO}_2$ ) goes through both subcritical and supercritical states, “a transcritical cycle”.

UVa



# Description of the CDTPC

- In the evaporator, it is obtained a better fit with the heat source when the heat is added to the working fluid in supercritical conditions.



# Modelling of the process

- Constant isentropic efficiencies of 75% are assumed for the pump as well as for the turbine.
- Steady state conditions.
- No pressure drop or heat loss in the evaporator, condenser or the pipes.
- An inlet temperature of the condensation water  $T_7=15$  °C.
- Working fluid condensation temperature  $T_3=25$  °C.
- Turbine inlet temperature  $T_1=150$  °C.
- Turbine inlet pressure  $P_1$ , varying from 66 bar until the net work was **around** zero.

# Modelling of the process.

- The cycle's total energy efficiency is:

$$\eta = \frac{\dot{W}_t - \dot{W}_p}{\dot{Q}_e}$$
$$\dot{W}_t = \dot{m} \times \eta_t \times (h_1 - h_2) \quad \dot{W}_p = \dot{m} \times (h_3 - h_4) / \eta_p$$
$$\dot{Q}_e = \dot{m} \times (h_1 - h_4)$$

- The exergy efficiency is defined as:

$$\eta_E = 1 - \left( \frac{\sum_i \dot{I}_i}{\sum_i \dot{E}_{in,i}} \right)$$

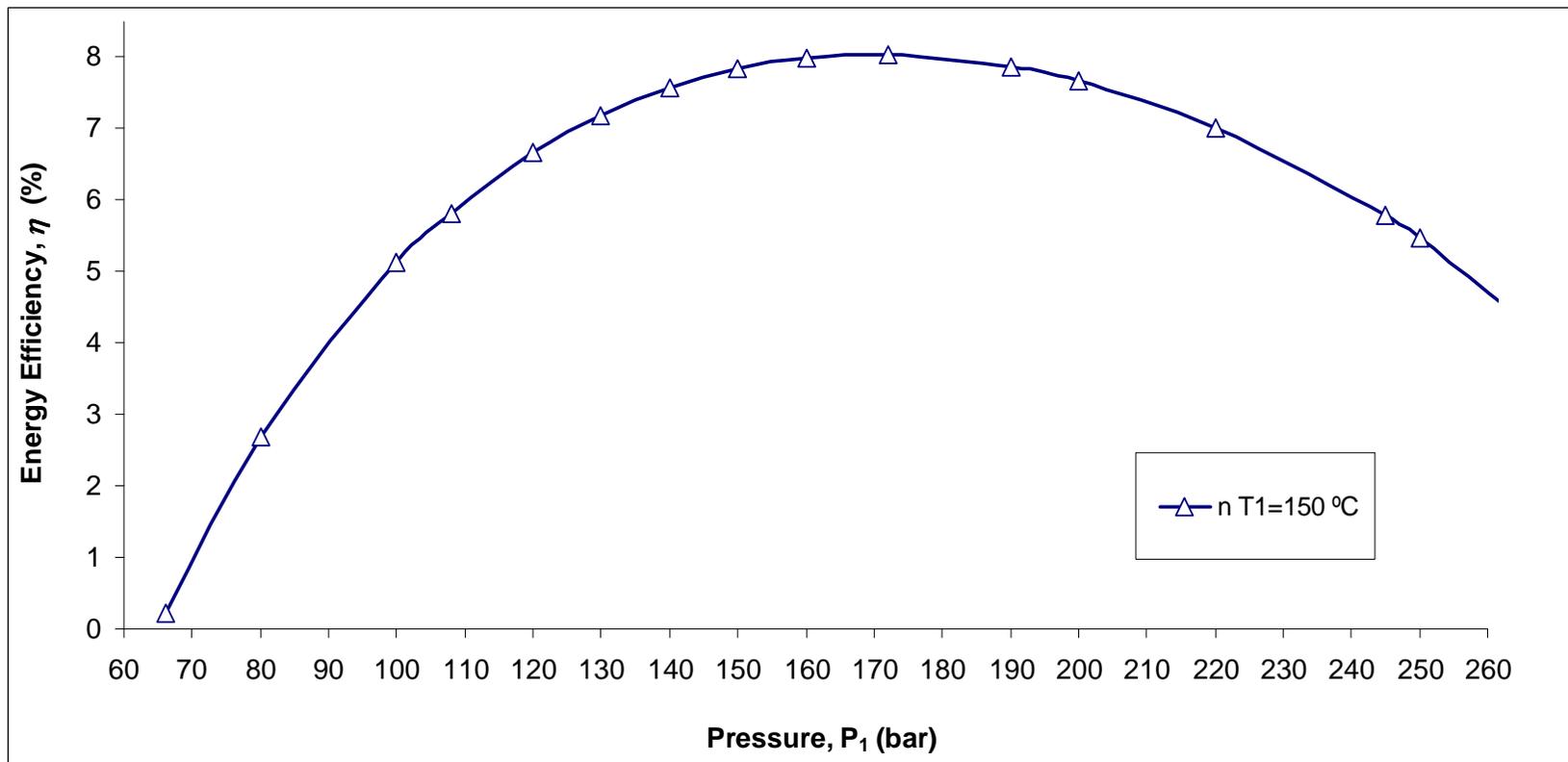
$\dot{I}_i$  is the exergy loss (destruction) of each component  $i$  (evaporator, turbine, condenser and pump) that can be found from an exergy balance.

$\dot{E}$  Exergy rate.

# Results and discussion

## *ENERGY ANALYSIS*

- Energy efficiency increases as pressure  $P_1$  rises.
- A parabola-like behaviour.

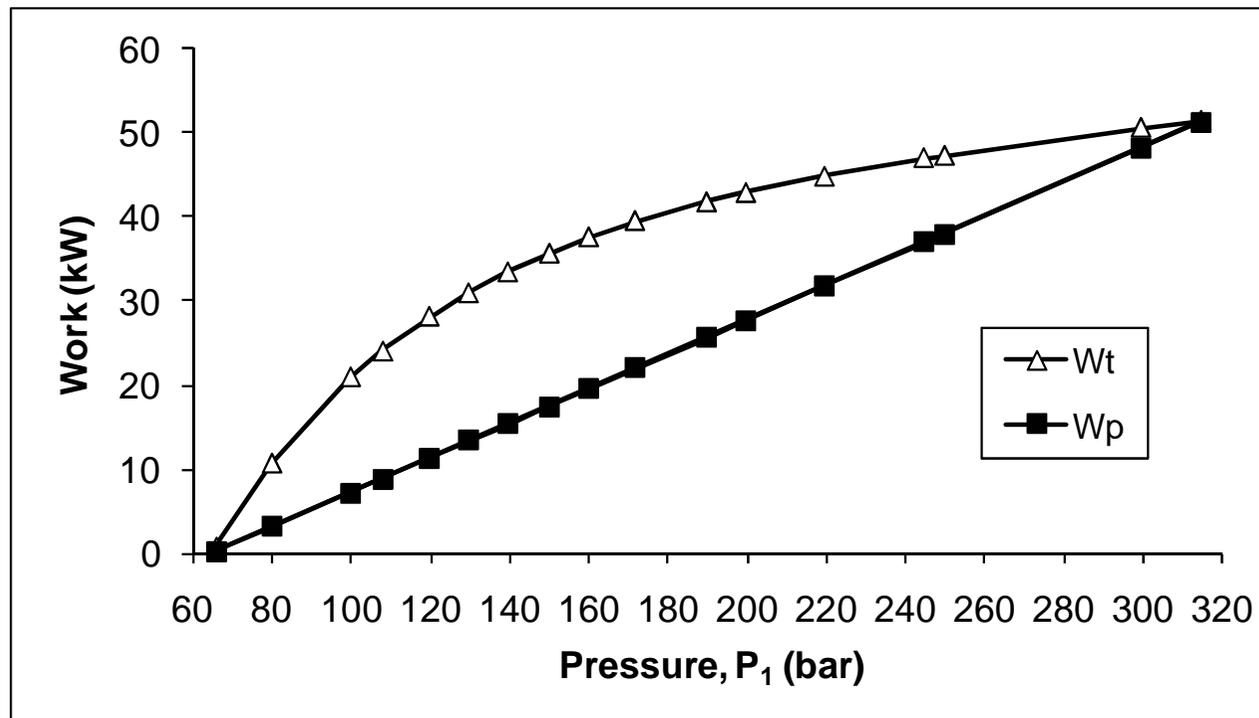


Energy efficiency vs Pressure  $P_1$  with an inlet temperature to the turbine at 150 °C.

# Results and discussion

## *ENERGY ANALYSIS*

- Energy.



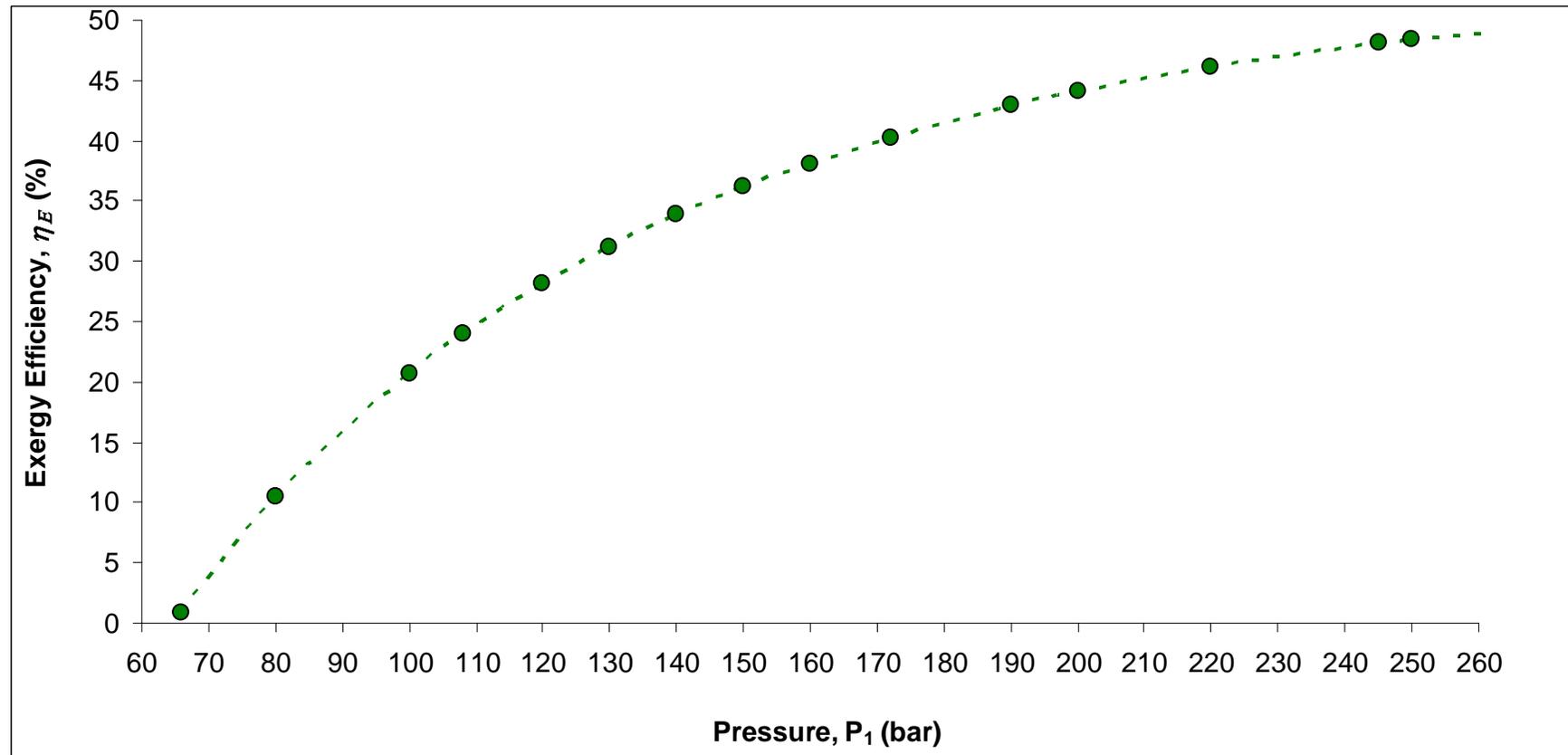
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**Work produced and consumed by the turbine and the pump, respectively vs Pressure  $P_1$  with an inlet temperature to the turbine at 150 °C.**

# Results and discussion

## *EXERGY ANALYSIS*

- Exergy.

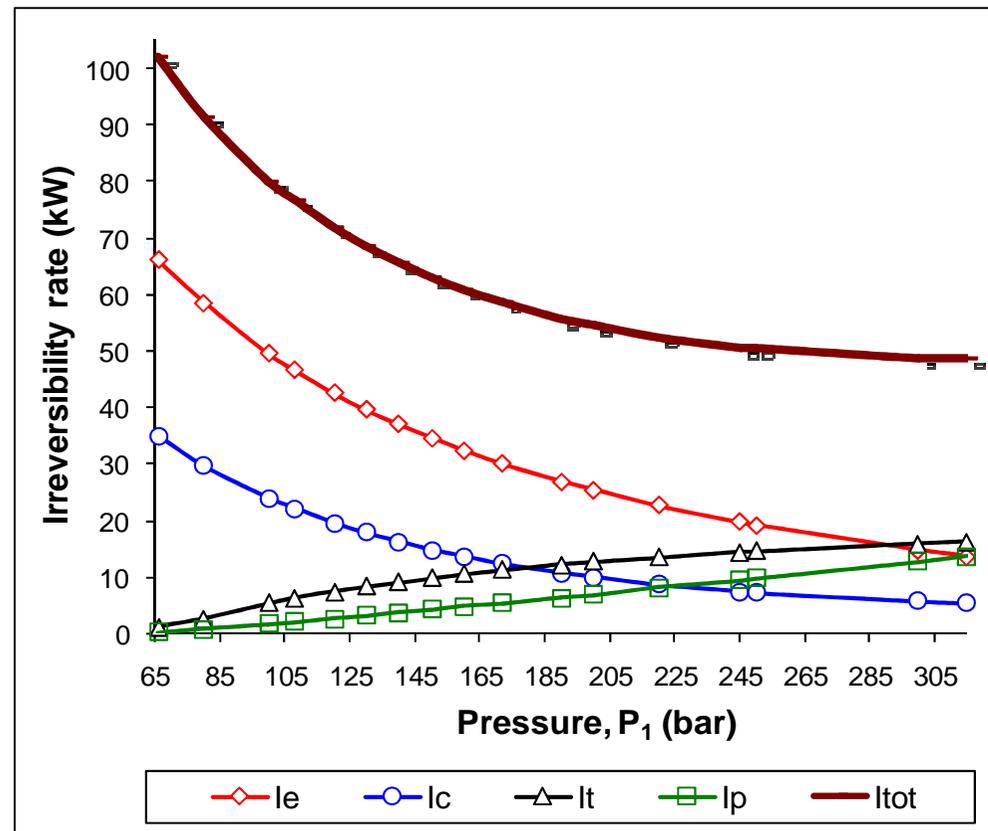


Exergy efficiency vs Pressure  $P_1$  with an inlet temperature to the turbine at 150 °C.

# Results and discussion

## EXERGY ANALYSIS

- Irreversibilities of the process in each device.



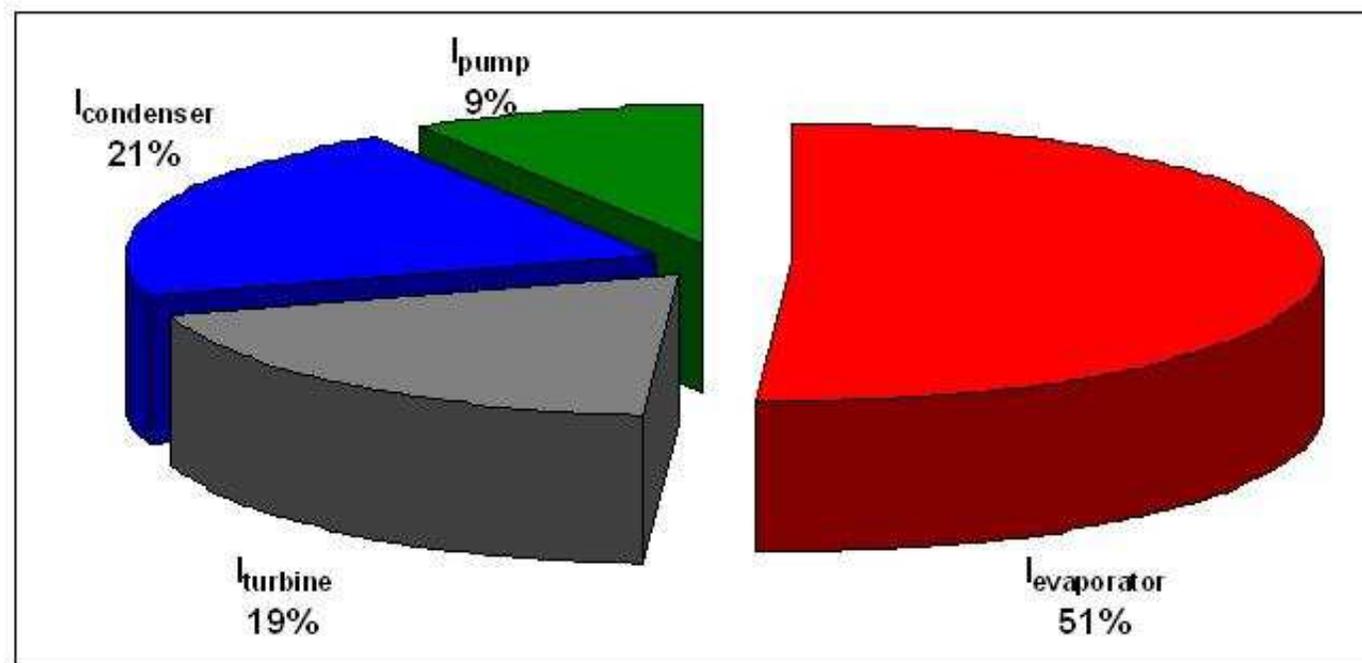
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- Effect of the inlet pressure to the turbine over the irreversibilities of the process with an inlet temperature to the turbine at 150 °C.

# Results and discussion

## *EXERGY ANALYSIS*

- Irreversibilities of the process in each device at optimum conditions.



Irreversibilities of the process in optimum conditions with an inlet temperature to the turbine at 150 °C.

# Conclusions

## Optimum conditions of design at $T_1=150^\circ\text{C}$ .

Parameter			
Pressure (bar)	$\eta$ (%)	$\eta_E$ (%)	$wne$ (kJ/kg)
172.0	8.0	40	17.5

- ✓ **CDTPC is suitable for the production of useful energy utilising low enthalpy heat.**
- ✓ **It is possible to operate with a CDTPC in relatively low temperature ranges.**
- ✓ **Maximum point in the energy efficiency of the process is found.**



Low temperature heat source for power generation:

Exhaustive analysis of a carbon dioxide transcritical power cycle

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**THANK YOU FOR YOU  
ATTENTION.**



**frevel@cartif.es**



**[www.cartif.es](http://www.cartif.es)**

**Parque Tecnológico de Boecillo, 205**

**47151 Boecillo, Valladolid. Spain.**

**Tel: +34 983 143 804**

**Fax: +34 983 546 521**